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2316

Class – B.Sc. IV Subject – Physics Paper – A (Quantum Mechanics)

Time Allowed : 3 Hours

Maximum Marks: 35

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The candidates are required to attempt ONE question each from Section B,C,D and E. Section A consisting of Seven short answer type questions is compulsory. All questions carry equal marks.

Section A

Q1. Attempt all parts:

- (i) Why is the wave nature of matter not more apparent to us in our daily observations?
- (ii) What do you understand by the term work function?
- (iii) What is the condition for orthogonal wave function?
- (iv) What is Raman effect
- (v) What is Auger effect?
- (vi) Explain the term degeneracy.
- (vii) What are eigen functions and eigen value?

Section - B

- Q 2.(a) Define phase velocity of a de broglie wave. Show that phase velocity can exceed the velocity of light in vacuum.
 - (b) A particle is described by a wave function

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 $\Psi = \sqrt{\frac{2}{L}} \sin \frac{n \pi x}{L}$ for 0 < n < L and = 0 else where.

Find expectation value of momentum of the particle.2.



By using Heisenberg's uncertainty principle, explain.

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- 1) Non- existence of electrons within nucleus.
- (2) Existence of finite zero point energy.
- (b) What are the conditions for a wave function to be called well behaved wave function? 2

Section - C

- Q 4. Define Hermitian operator. Prove that Hermitian operator has real Eigen value. Define orthogonal wave functions. Prove that two Eigen functions of hermitian operator beloncing to different Eigen values are orthogonal.
- Q 5. State and explain Ehrenfests theorem on the motion of wave pockets. 7

Section D

- Q 6. Solve schrodinger equation for one dimensional harmonic oscillator to obtain its energy levels. What is zero point energy of Harmonic oscillation Explain.
- **Q7.** A particle of mass M and total energy E moves from the region of zero potential to the region of constant potential V_0 . Given that $E > V_0$ derive expressions for reflection coeff R and transmission Coeff T. Prove that R + T = 1. 7

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Section E

- Q8. Give the origin of continous and characteristic X ray spectra. Why is characteristic spectra so called? 7
 - Q9. Obtain an expression for rotational energy levels in diatomic molecule and frequency rotational spectrum. State clearly the selection rules. Show that protational spectra the frequencies are equally spaced. 7

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